### Not all turtles are slow



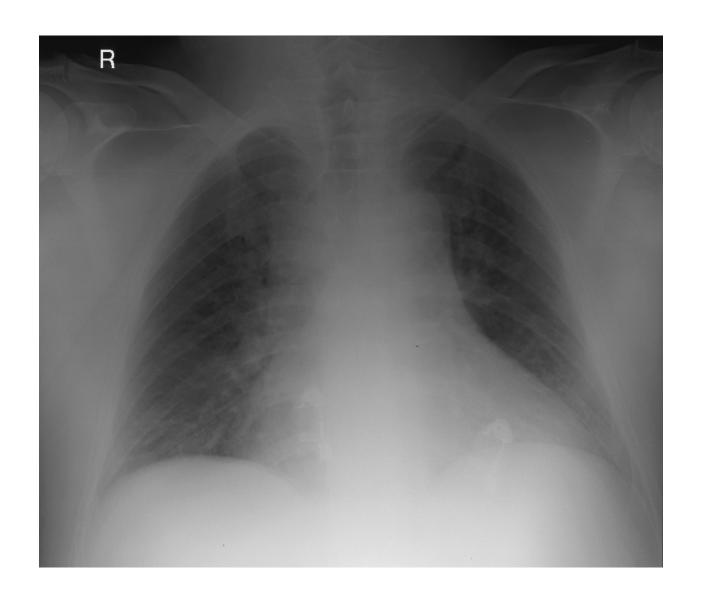
Chrysemys picta marginata; the diving turtle

### 47-year-old, shortness of breath and leg swelling for 2 weeks



- Unemployed Berliner, garden colony (Laubenpieper)
- No meds, "denies" alcohol intake
- BP 115/60, HR 96/Min, RR 36/Min
- Rales bilateral
- Distant heart sounds, no murmurs, ?S3
- Bowel sounds diminished
- Edema to sacrum
- Suspect CHF

### Portable films are next to worthless



### Acute labs

- Hb 13 g/dl, Hematocrit 38 vol%
- pH 7.30, PaCO<sub>2</sub> 20, PaO<sub>2</sub> 105 (mm Hg), HCO<sub>3</sub> 10,
- Na 123, K 6.9, Cl 87 (mmol/l)
- Glucose 5.6 mmol/l
- Lactate 14 mmol/l
- Anything life threatening here?

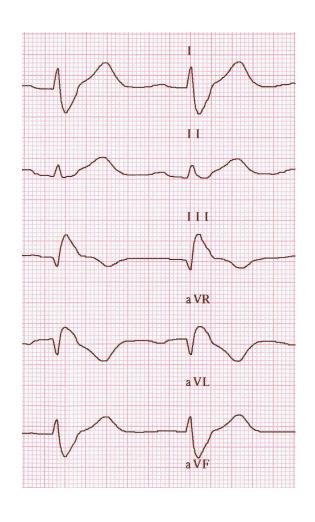
### Anything lethal here?

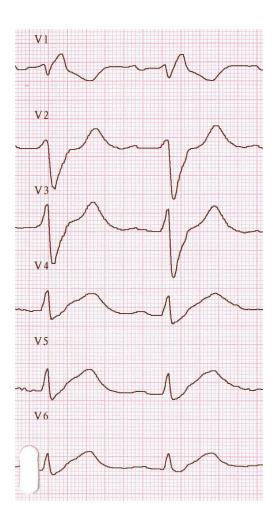
- Acid-base problem?
- Hyperkalemia?
- Hyponatremia?
- Poor oxygen delivery?
- Heart failure?
- Just being in our ICU?

### The answer is "jawohl"!

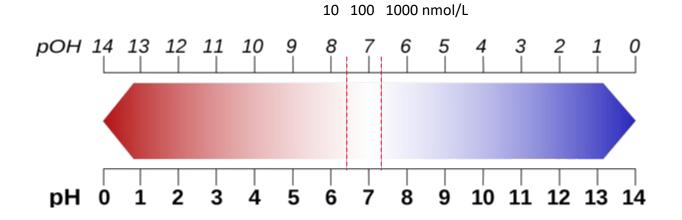
- Alveolar Gas Formula (Sea level)
- $PAO_2 = 150 (PaCO_2)/0.8 [RQ]$
- $\bullet$  = 150 105 + 25 = 20 mm Hg
- A-a gradient modestly elevated (not a problem).
- What is the hydrogen ion concentration at pH 7.30?
- How many people die because of hydrogen ions?
- None!

### How about Hyperkalemia?

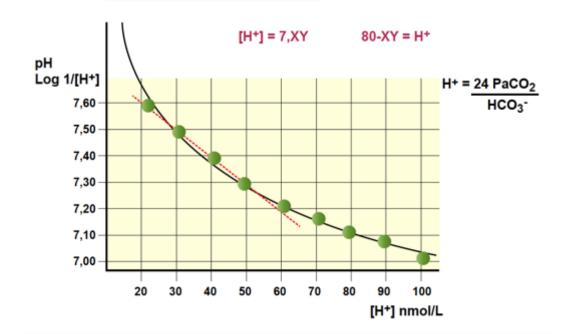




### The pH was 7.30. What is the [H<sup>+</sup>] in nanomols per liter



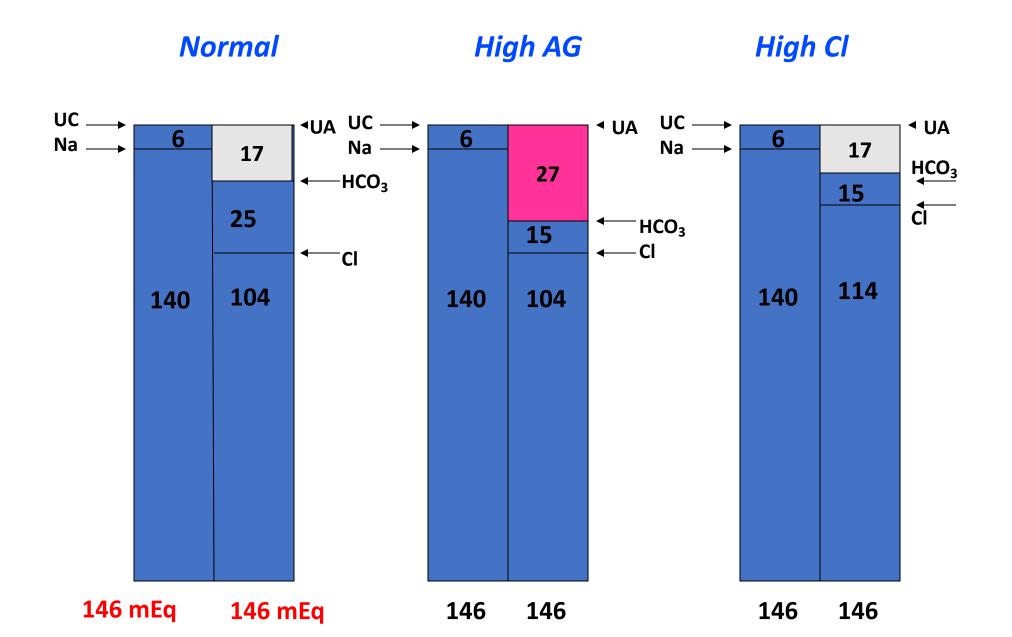
"PUISSANCE" HYDROGEN Sorensen & Hasselbalch



### Now the "nitty-gritty"!

- Is there an acid-base problem?
- Acidemia with a low HCO<sub>3</sub> and low PaCO<sub>2</sub>
- Metabolic Acidosis
- HCO<sub>3</sub> fell by 15 mmol (25 to 10)
- Decrease in PaCO<sub>2</sub> was 20 mm Hg
- Metabolic Acidosis, slightly overcompensated ie. respiratory Alkalosis

### Metabolic acidosis is caused by gain of $[H^+]$ or loss of $[HCO_3^-]$



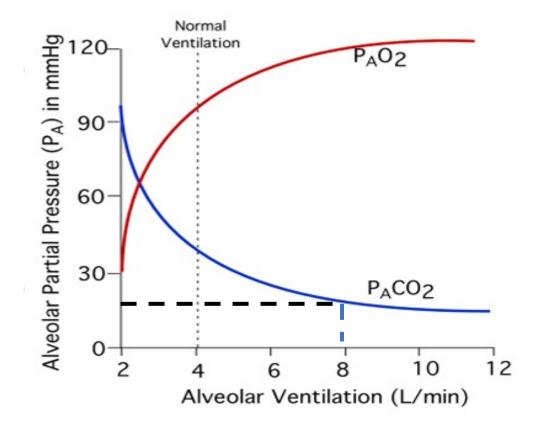
### Laubenpieper has an elevated AG

- AG = Na CI  $HCO_3 = 26 (12\pm 2)$
- AG is 14 mmol/l too high
- Lactate is elevated by 14 mmol/l
- Ahah! For every mmol/l increase in lactate we observe one mmol/l increase in AG
- "Delta" for "Delta"
- So we have a pure anion-gap metabolic acidosis!

### Is the respiratory compensation appropriate?

 $H = (24 \times PCO_2)/HCO_3$ 

 $PAO_2 = 150-PaO_2+PaCO_2/0.8$ 



### Laubenpieper is shipped up to our ICU

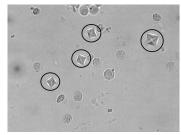
- Leukocytes 16 000/μ<sup>3</sup>
- D Dimer 0.48 mg/dl, Trop T 0.05 μg/l
- aPTT 18 sec, ALAT, ASAT, Bilirubin slightly elevated
- Crea 243 µmol/l, Urea 18 mmol/l
- Osmo 287 mosm/l
- Ethanol not detectable

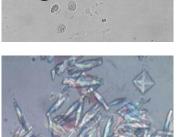
### Can we dismiss methanol and ethylene glycol poisoning?

- Calculated Osmo = 2(Na+K) + Glu + Urea
- $\bullet$  = 2(129) + 5.6 + 19 = 283
- Difference is only 4 mosm
- Most common cause of increased AG today?

### Mnemonics: Kussmaul or Goldmark?)

- Methanol; uremia; diabetic ketoacidosis (DKA); paraldehyde, phenformin; iron, isoniazid; lactic; ethylene glycol; salicylates (English)
- Kussmaul: Ketoazidose, Urämie, Salizyl-Säure, Metanol, Äthylenglykol, (mehr) Urämie, Laktat (German)
- Goldmark: Glycol, Oxoproline, L-Lactate, D-Lactate, Methanol, Aspirin, Renal Failure, Ketones







### Can his urine tell us anything?

Sediment for ethylene glycol

 Ferric chloride for acetyl salicylic acid, ketonuria, urea

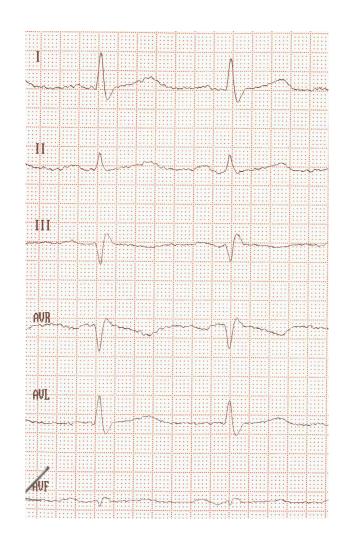
 Lactate is 100% absorbed and does not appear in urine

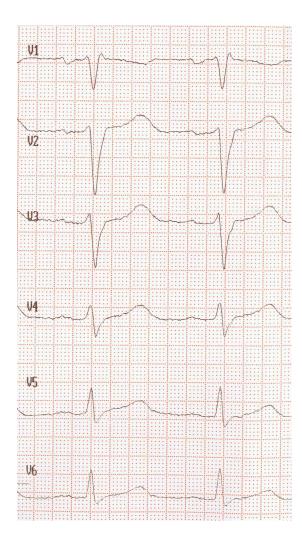
UK, UNa and UCreatinine

### We check out his urine!

- UNa 10 mmol/l
- UCI 15 mmol/l
- UK 35 mmol/l
- Sp.Grav. 1.030 (What is Uosm??)
- Prot 2+
- WBCs und granulated cylinders
- Ketone not present
- Mnemonics lead us to lactate
- Prerenal acute renal failure?

# Patient stayed oliguric despite a truck load of furosemide. Hemodialysis was done and fixed his ECG.

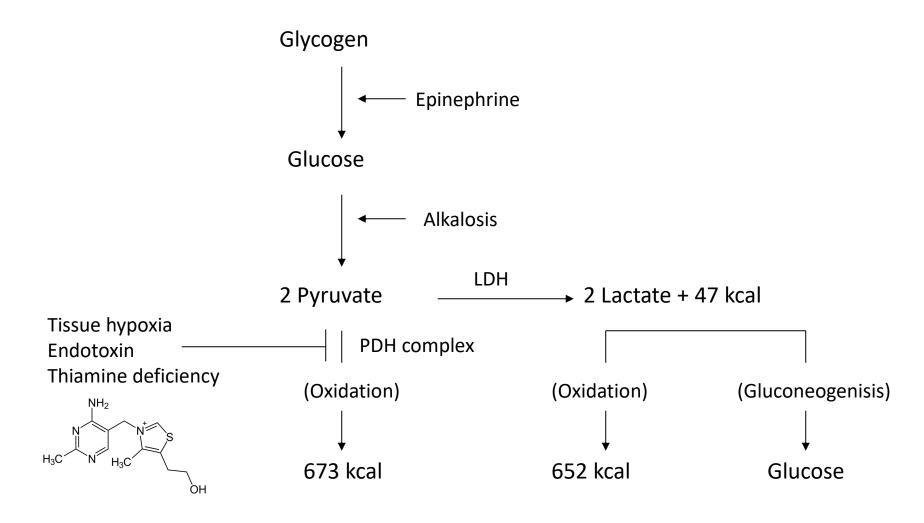




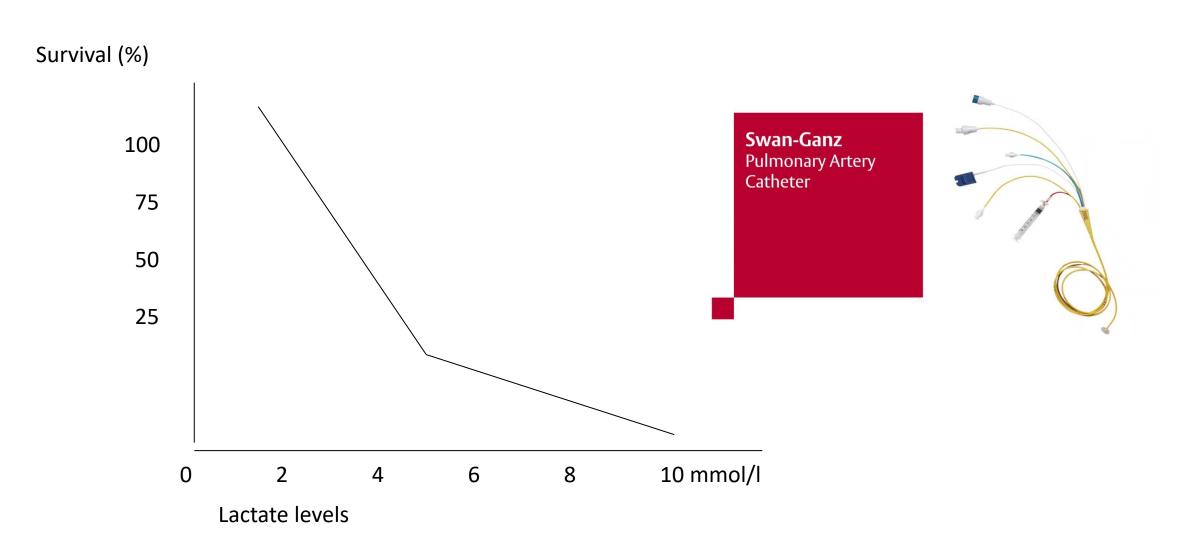
### Glukose+2ATP+2H<sub>2</sub>PO<sub>4</sub> -> 2Lactate+2ADP+2H<sub>2</sub>O

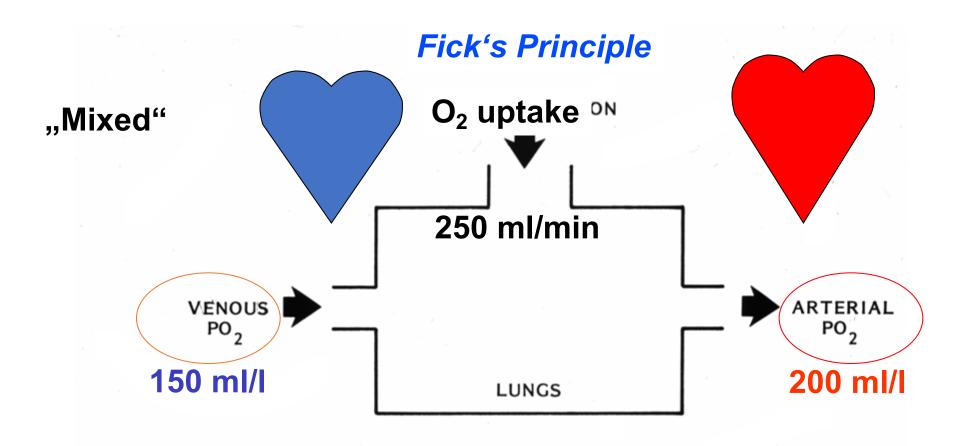
- Lactate is the end product of anaerobic glycolysis
- Production is 1 mmol/kg/h, circa 2000 mmol/day for our 80 kg Laubenpieper
- Production in skeletal muscle, GI tract, brain and erythrocytes
- The liver uses it (for gluconeogenesis) or for fuel

## **Glycolysis**



## Tissue hypoxia - lactate associated with shock (Circulation 1975)









C is "content" i.e. Oxygen content

### But Laubenpieper was not in Shock!

- CVP 25 mm Hg
- PCWP 20 mm Hg
- CO 9 I/min
- Peripheral resistance 336 dyne/cm<sup>-5</sup>
- What was DO<sub>2</sub>, consumption VO<sub>2</sub> and O<sub>2</sub> extraction?
- No Swan-Ganz catheters, unless you can do this!

$$R=rac{\Delta p}{\dot{V}}$$

### High output heart failure

- Hb(13 g/dl) x 1.34 = 17.4 ml  $O_2$ /dl or 174 ml/l
- CO = 9 I/min
- $DO_2 = 174x9 = 1566 \text{ ml/min } (O_2 \text{ offered per min.})$
- $VO_2 = DO_2 \times (SaO_2 SvO_2)$
- $SaO_2 = 100\%$
- $PvO_2$  was 60 mmHg;  $SvO_2$  = 90% (40 mm Hg; 75%)
- $VO_2 = 156.6 \text{ ml/min}$
- $O_2ER = 10\%$  (normal  $O_2ER$  is 25%)
- Why does the Laubenpieper not utilize O<sub>2</sub>?

### What leads to O<sub>2</sub> uncoupling?

- Poisoning cytochrome C oxidase
  - Cyanide, CO
- Dinitrophenol
  - Atractyloside (a plant glycoside)
  - Bongkrekic
- Metformin
- Disturbances of PDH
- Endotoxin
- Thiamine deficiency

### Who first described Beriberi?





J.UGDUNI BATAVORUM apud GEORGIUM WISHOFF.

Nicholas Tulp aus Leyden, 1739 Rembrandt in the "Anatomy demonstration"

Christiaan Eijkmann, polished rice, Nobel Prize 1929

### Laubenpieper received Thiamine

- Within 12 h Lactate fell to 1 mmol/l
- pH rose to 7.4
- HCO<sub>3</sub> increased to 24 mmol/l
- O<sub>2</sub> extraction was 20%
- Cardiac function improved within 1 week
- Liver and renal function normalized
- Signed out against medical advice



# Plasma values (mmol/l) of Chrysemys picta marginata

#### Prior to hibernation

• Na 117

• Cl 73

• HCO<sub>3</sub> 39

• AG 5

Lactate4

• K 2

• Mg 2.6

• Ca 3.7

### Stuck under water for 5 months

• Na<sup>+</sup> 99

• Cl<sup>-</sup> 44

•  $HCO_3^-$  5  $\Delta 34$ 

• AG<sup>-</sup> 50 ∆45

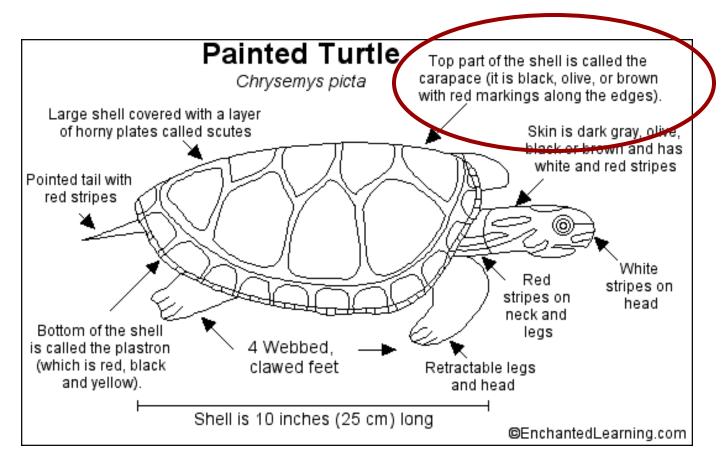
• Lactate<sup>-</sup> 185 <u>∆</u>181

• K<sup>+</sup> 10

• Mg<sup>2+</sup> 12

• Ca<sup>2+</sup> 59

### A physiological lactic acidosis



HCO<sub>3</sub> decease is much lower than would be expected than the lactate increase. Because H<sup>+</sup> Ions were buffered with CaCO<sub>3</sub> in the turtle's shell.

### Teaching points

- Acid-base problems are not acutely lethal.
- Hyperkalemia (when treated) needs insulin (and glucose)
- Swan-Ganz allows metabolic assessments: DO<sub>2</sub> and VO<sub>2</sub>.
- AG (like mosm/L) is a clinical tool, not a law of nature
- Lactic acidosis is today the most common cause of AG acidosis – there are two forms